



**Chemistry 0010 Course Outline
Introductory Chemistry
2016-17**

Instructor: Jan Mathers

Lab Instructor: Dr. Christina Booker

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Office Hours: extra help will be available on Friday mornings. The time will depend on student schedules.

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Course Description

Chemistry 0010 is equivalent to the Ontario Grade 12U level chemistry. Some topics from Grade 11U are also covered. Topics to be studied include; atomic theory, molecular structure, types of chemical reactions and stoichiometry, thermochemistry, kinetics, equilibrium, oxidation and reduction reactions and organic chemistry. Laboratory work will be an important component of this course.

Course Times

The schedule for lectures, tutorials, labs and midterms will be handed out in class.

Monday classes (lectures, tutorials or midterms) 2:30 - 4:30 p.m. St. James 202

Thursday classes (lectures, tutorials or midterms) 2:30 - 4:30 p.m. St. James 202

Tuesday labs: 2:30 pm. and 4:30 pm. in MRW 156. Each lab session is two hours long.

Required Materials All are available at the Western Bookstore

Chemistry 0010 Notes 2016-17

Chemistry 0010 Lab Manual 2016-17

Lab Coat and scientific calculator

The Notes and Lab Manual are Custom Course Materials. You may have to purchase a voucher and pick up the material the following day.

Course Evaluation

Lab Reports (7 + Introductory Lab)	15 %	
Tutorial Quizzes (6)	15 %	
Midterm Test (November 17)	20 %	(tentative date)
Midterm Test (January 26)	20 %	(tentative date)
Final Exam (scheduled by the Registrar)	30 %	

You must pass the laboratory portion of the course to pass the course (mark $\geq 7.5 / 15$).

All the labs and all the tutorial quizzes count toward the mark. None are 'dropped'.

Laboratory Requirements

A lab coat is mandatory and may be purchased at the Western bookstore. Safety Glasses will be provided.

Dress Code: A lab coat, long pants (must come to the ankles), socks that cover the ankles, and closed shoes are mandatory, hair must be tied back, if possible. Students will be asked to leave the lab if these requirements are not met and a mark of zero will be assigned for that lab.

Attendance: All labs count toward the lab mark. **Any student who is more than 5 minutes late will not be permitted to do the experiment.** The clock in the lab will be used as the basis for this decision. Please contact academic counseling if you miss a lab due to illness or other extenuating circumstances.

Objectives: For many of the experiments, students will work with a partner. Good communication will likely result in a more successful experiment. At the completion of the course, the student should be able to: correctly use lab equipment such as burettes, balances and pipettes; titrate a solution to the endpoint; follow written instructions; record pertinent observations; use experimental data to complete calculations using the correct number of significant figures; compare their data to 'real' data and make constructive suggestions as to why there may be differences.

Tutorials

There will be six, in-class, tutorial tests throughout the year. The first 40 minutes of the class will be an opportunity to ask questions and get help, followed by a quiz. Please contact academic counseling if you miss a tutorial test due to illness or other extenuating circumstances.

Contacting Students

Students will receive email at their UWO address only. Emails sent to the entire class will be through OWL. Please check for email on a regular basis.

Absences

Absence from any activity that is marked (labs, tutorial tests and midterms) must be dealt with by academic counseling. Documentation will be required.

Department Policy: For academic accommodation to be considered for any course component worth less than 10% of the final course grade, it is the responsibility of the student to approach the course instructor(s) in a timely fashion. Documentation may be required to be submitted to the academic advisor. If documentation is required, the request for accommodation will be decided by the academic advisor in consultation with the instructor. If documentation is not required, the instructor will make the final decision. The policies governing requests for academic accommodation for course components worth 10% or more of the course grade are outlined in the Academic Policies section included at the end of the course outline.

Late Assignments

Lab reports are marked out 10. Any that are handed in *late* will have *one mark per day* deducted and will not be accepted once the marked reports have been returned to the other students, unless documentation is received from academic counselling. Lab reports are usually due on the Thursday following the Tuesday lab period.

Code of Conduct

All classes will start promptly. Please bring the course notes and a calculator to class. It is expected that you will come to class prepared to listen and ask questions.

Some Suggestions

If you are having trouble with any of the concepts, contact the instructor and ask for help as soon as possible. Attendance is strongly correlated to the mark achieved in this course. Math skills are also quite important.

Some additional classes will be available for those that need more instruction with math. An assessment will be given to evaluate your math skills and does not count toward the course mark.

It is difficult to do well in this course without doing the problems on your own. There are lots of questions in the Course Notes. The final answers to most of the questions are given in the course notes and the **full solutions are on course website**. Do not just read the solutions. Do as many as you need to grasp the concept. Practice exams will be provided.

Topics And Objectives

It is expected that students will develop appropriate study skills for their learning style and make use of the resources available to them. This is a problem solving course. While there are many examples and questions in the course notes, students are expected to be able to work through problems that are not identical to ones they have already been exposed to. There are many different topics in this course. However, as we proceed through them, students should be able to connect the various pieces of information to allow greater understanding of each topic.

The *objectives* listed with each topic will not cover every type of question that will be asked. For each topic the student should be able to:

Chemistry: the Study of Matter

identify the major parts of an atom; distinguish between chemical and physical properties;
name simple compounds

The Mole

solve calculations involving Avogadro's number, moles, molar mass, molarity, and the composition of compounds or mixtures

Chemical Reactions and Stoichiometry

write and balance specific types of reactions and carry out calculations, using stoichiometry, involving amounts of products made and reactants used; determine oxidation states, solubility, and pH; determine the limiting reagent and yield

Atomic Theory

describe the historical development of the structure of the atom and arrangement of the electrons; use conventional methods to describe the arrangement of the electrons in atoms and ions (electron configuration, orbital box notation, quantum numbers)

Periodic Properties

use core charge and the distance of an electron from the nucleus to explain the trends in the periodic table with respect to atomic size, ionization energy, electron affinity and electronegativity

Structures and Bonding

describe the types of bonding that occurs between elements from different parts of the periodic table: metallic, ionic and covalent

Lewis Structures

apply the rules for Lewis structures to draw molecules with covalent bonds and determine some features of the compound such as resonance and bond order

VSEPR Rules for Molecular Shapes

apply the VSEPR theory rules to determine the shape of a compound at a central atom, the hybridization of the central atom and the polarity of the compound

Intermolecular forces

determine the type and relative strength of the forces that exist between molecules based on their structure

Ideal Gases

use the Ideal gas law to carry out calculations involving one gas or a mixture of non-reacting gases; identify the forces that account for differences between an ideal and real gas

Thermodynamics

calculate the energy changes involved in various processes such as heating, cooling, changes of state and chemical reactions; distinguish between energy change and enthalpy change; use Hess's Law to determine the enthalpy change for a reaction; write formation reactions; use calorimetry to determine the heat of a reaction; describe entropy and Gibbs Free Energy

Kinetics

use stoichiometry to describe the relationship between the rates of reaction of the various components of a reaction; use Collision Theory to explain the factors affecting the rate of a reaction; given the appropriate information determine the rate law for a specific reaction; identify the order of a reaction, the intermediates and any catalysts; calculate the half life time or concentrations involved in first order reactions; carry out calculations using the Arrhenius equation

Chemical Equilibrium

write the equilibrium constant expression for a given reaction and carry out simple calculations; use Le Chatelier's Principle or the reaction quotient to determine the direction in which a reaction will proceed; use various strategies to simplify calculations involving equilibrium constants; write the equilibrium reaction and carry out calculations for equilibrium situations involving low soluble salts, weak acids, weak bases, salts or buffer solutions; identify various titration curves, choose an appropriate indicator for a titration

Redox Reactions

determine the oxidation state of an element; balance a redox reaction and identify the species being oxidized or reduced

Electrochemistry

identify the parts of an electrochemical cell; write cell notations; use the cell potential to determine the spontaneity of the reaction; use Faraday's Law to calculate various parameters such as current, time or the amount of product in an electrolytic process

Organic Chemistry and Polymers

identify various organic functional and family groups; name simple compounds; identify conformers, structural isomers and geometric isomers; describe properties relative to other family groups; identify several common types of organic reactions, draw/name the product of the reactions of several types of organic compounds, identify the two types of reactions used to make polymers; given the starting materials draw the polymer that could be made, identify the common polymers

2016-17 BRESCIA UNIVERSITY COLLEGE ACADEMIC POLICIES AND REGULATIONS

1. POLICY REGARDING MAKEUP EXAMS AND EXTENSIONS OF DEADLINES

When a student requests academic accommodation (e.g., extension of a deadline, a makeup exam) for work representing 10% or more of the student's overall grade in the course, it is the responsibility of the student to provide acceptable documentation to support a medical or compassionate claim. All such requests for academic accommodation **must** be made through an Academic Advisor and include supporting documentation. Academic accommodation for illness will be granted only if the documentation indicates that the onset, duration and severity of the illness are such that the student could not reasonably be expected to complete her academic responsibilities. Minor ailments typically treated by over-the-counter medications will not normally be accommodated. Documentation shall be submitted as soon as possible to the student's Academic Advisor indicating the period of illness and when the student should be able to resume academic responsibilities. Students must submit their documentation along with a request for relief specifying the nature of the accommodation being requested no later than two business days after the date specified for resuming responsibilities. Appropriate academic accommodation will be determined by the Dean's Office in consultation with the student's instructor(s). Please note that personal commitments (e.g., vacation flight bookings, work schedule) which conflict with a scheduled test, exam or course requirement are **not** grounds for academic accommodation.

A UWO Student Medical Certificate (SMC) is **required** if a student is seeking academic accommodation on medical grounds. This documentation should be obtained at the time of the initial consultation with the physician/nurse practitioner or walk-in clinic. A SMC can be downloaded from: <http://www.westerncalendar.uwo.ca/2016/pg117.html> The student must request documentation sufficient to demonstrate that her ability to meet academic responsibilities was seriously affected. Please note that under University Senate regulations documentation stating simply that the student "was seen for a medical reason" or "was ill" is **not** adequate to support a request for academic accommodation.

Whenever possible, requests for academic accommodation should be initiated in advance of due dates, examination dates, etc. Students must follow up with their professors and Academic Advisor in a timely manner.

The full policy on requesting accommodation due to illness can be viewed at:
http://www.uwo.ca/univsec/pdf/academic_policies/appeals/accommodation_illness.pdf

2. ACADEMIC CONCERNS

If you feel that you have a medical or personal problem that is interfering with your work, contact your instructor and Academic Advisor as soon as possible. Problems may then be documented and possible arrangements to assist you can be discussed at the time of occurrence rather than on a retroactive basis. Retroactive requests for academic accommodation on medical or compassionate grounds may not be considered.

If you think that you are too far behind to catch up or that your work load is not manageable, you should consult an Academic Advisor. If you consider reducing your workload by dropping one or more courses, this must be done by the appropriate deadlines (refer to the Registrar's website, www.registrar.uwo.ca, for official dates). You should consult with the course instructor and the Academic Advisor who can help you consider alternatives to dropping one or more courses. *Note that dropping a course may affect OSAP eligibility and/or Entrance Scholarship eligibility.*

The Dean may refuse permission to write the final examination in a course if the student has failed to maintain satisfactory academic standing throughout the year or for too frequent absence from the class or laboratory (<http://www.westerncalendar.uwo.ca/2016/pg130.html>)

3. ABSENCES

Short Absences: If you miss a class due to a minor illness or other problems, check your course outline for information regarding attendance requirements and make sure you are not missing a test or assignment. Cover any readings and arrange to borrow notes from a classmate. Contact the course instructor if you have any questions.

Extended Absences: If you have an extended absence, you should contact the course instructor and an Academic Advisor. Your course instructor and Academic Advisor can discuss ways for you to catch up on missed work and arrange academic accommodations, if appropriate.

4. POLICY ON CHEATING & ACADEMIC MISCONDUCT

Scholastic offences are taken seriously and students are directed to read the appropriate policy, specifically, the definition of what constitutes a Scholastic Offence, at the following Web site: <http://www.westerncalendar.uwo.ca/2016/pg113.html>

Students are responsible for understanding the nature of and avoiding the occurrence of plagiarism and other academic offences. Students are urged to read the section on Scholastic Offences in the Academic Calendar. Note that such offences include plagiarism, cheating on an examination, submitting false or fraudulent assignments or credentials, impersonating a candidate, or submitting for credit in any course without the knowledge and approval of the instructor to whom it is submitted, any academic work for which credit has previously been obtained or is being sought in another course in the University or elsewhere. Students are advised to consult the section on Academic Misconduct in the Western Academic Calendar.

If you are in doubt about whether what you are doing is inappropriate or not, consult your instructor, the Student Services Centre, or the Registrar. A claim that "you didn't know it was wrong" is not accepted as an excuse.

The penalties for a student guilty of a scholastic offence (including plagiarism) include refusal of a passing grade in the assignment, refusal of a passing grade in the course, suspension from the University, and expulsion from the University.

Plagiarism:

Students must write their essays and assignments in their own words. Whenever students take an idea or a passage from another author, they must acknowledge their debt both by using quotation marks where appropriate and by proper referencing such as footnotes or citations. Plagiarism is a major academic offence (see Scholastic Offence Policy in the Western Academic Calendar).

All required papers may be subject to submission for textual similarity review to the commercial plagiarism detection software under license to the University for the detection of plagiarism. All papers submitted for such checking will be included as source documents in the reference database for the purpose of detecting plagiarism of papers subsequently submitted to the system. Use of the service is subject to the licensing agreement, currently between The University of Western Ontario and Turnitin.com.

Computer-marked Tests/exams:

Computer-marked multiple-choice tests and/or exams may be subject to submission for similarity review by software that will check for unusual coincidences in answer patterns that may indicate cheating. Software currently in use to score computer-marked multiple-choice tests and exams performs a similarity review as part of standard exam analysis.

5. PROCEDURES FOR APPEALING ACADEMIC EVALUATIONS

All appeals of a grade must be directed first to the course instructor. If the student is not satisfied with the decision of the course instructor, a written appeal signed by the student must be sent to the Department Chair. If the response of the department is considered unsatisfactory to the student, she may then submit a signed, written appeal to the Office of the Dean. Only after receiving a final decision

from the Dean may a student appeal to the Senate Review Board Academic. A Guide to Appeals is available from the Ombudsperson's Office, or you can consult an Academic Advisor. Students are advised to consult the section on Student Academic Appeals under Academic Rights and Responsibilities in the Western Academic Calendar (<http://www.westerncalendar.uwo.ca/2016/pg112.html>)

6. PREREQUISITES AND ANTIREQUISITES

Unless you have either the prerequisites for a course or written special permission from your Dean to enroll in it, you will be removed from the course and it will be deleted from your record. This decision may not be appealed. You will receive no adjustment to your fees in the event that you are dropped from a course for failing to have the necessary prerequisite(s).

Similarly, you will also be deleted from a class list if you have previously taken an antirequisite course unless this has the approval of the Dean. These decisions may not be appealed. You will receive no adjustment to your fees in the event that you are dropped from a course because you have taken an antirequisite course.

7. SUPPORT SERVICES

The Brescia University College Registrar's website, with a link to Academic Advisors, is at <http://brescia.uwo.ca/academics/registrar-services/> . The website for the Student Development Centre at Western is at <http://www.sdc.uwo.ca/> . Students who are in emotional/mental distress should refer to Mental Health @ Western http://uwo.ca/health/mental_wellbeing/ for information including a complete list of options about how to obtain help.

Portions of this document were taken from the Academic Calendar, the Handbook of Academic and Scholarship Policy and the Academic Handbook of Senate Regulations. This document is a summary of relevant regulations and does not supersede the academic policies and regulations of the Senate of the University of Western Ontario.